



2010-04-01

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Recommended Citation

Liao, K. et al. (2010) Optical Limiting Study of Double Wall Carbon Nanotube-Fullerene Hybrids. *Chemical Physics Letters*, Vol. 489, Issues 4-6, Pages 207-211. 9 April. doi:10.1016/j.cplett.2010.03.010

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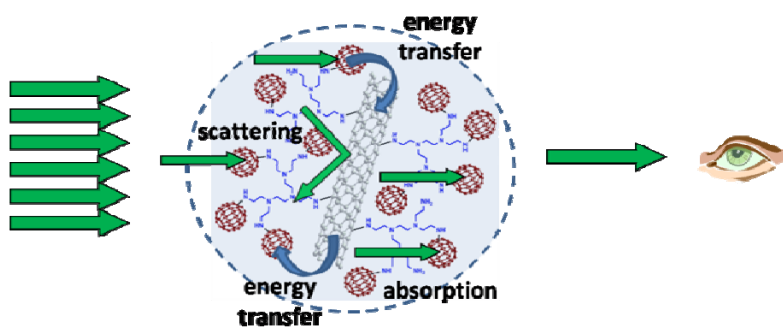
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Graphical Abstract

The double wall carbon nanotube-Fullerene hybrids exhibit superior optical limiting performance to those of Fullerenes and CNTs.



Abstract

In order to merge complementary temporal and spatial nonlinear optical characteristics of Fullerene and carbon nanotubes, synthesis of double wall carbon nanotube-Fullerene hybrid was performed by covalently linking DWNT and C₆₀ by amination reaction with polyethylenimine. DWNT-Fullerene hybrids were characterized by thermogravimetric analysis, UV-vis spectroscopy and transmission electron microscopy. Optical limiting performance of DWNT-Fullerene hybrids is superior to those of Fullerenes and SWNTs at the same level (~ 80%) of transmission. Whereas nonlinear scattering is an evident mechanism, reverse saturable absorption from Fullerene moieties has significant contribution. Charge transfer between the DWNT and Fullerene moieties may play an important role of optical limiting.

1. Introduction

Recently, there has been increasing interest in the application of nanotechnology to protective coatings, including polymer nanotubular thin film composites. The use of carbon nanotubes in nanocomposites revealed a multiplicity of applications as nano-fillers in polymer hosts including thermal sensors [1] and antistatic coatings [2]. However, the success of nanocomposite fabrication is still dependent on developing reliable synthetic methods in forming their primary building blocks, a process that is challenging at the nanoscale for both inorganic and organic nanomaterials. Nonetheless, there have been significant advances in the use of nanocomposites in various optical applications, including organic light emitting diodes [3] and optical limiters [4]. Since the invention of lasers in the 1960s, not only have they become powerful instruments for material assessment, but are also commonly used in our daily life, such as surgery and telecom applications. Protection from lasers is consequently not a trivial matter but is one of genuine concern from a public safety and technological perspective. The development of optical limiting materials/devices provides a valuable solution to the dangers of lasers, as well as various other forms of optical instruments being used. A successful optical limiter should strongly attenuate intense, potentially dangerous laser beams, while exhibiting high transmittance for low intensity ambient light. Investigating optical limiting materials/mechanisms/devices is therefore a very important area of research [5-8]. In the forefront of this field is a quest for ideal materials that include carbon nanomaterials [7,8], phthalocyanines [9,10], porphyrins [11], organometallics, inorganic semiconductors and liquid crystals [5,6]. In particular, the carbon nanomaterials: carbon black [12], fullerenes [13,14], carbon nanotubes (CNTs) [4,15-17] and graphenes [18] all make a great contribution to laser protection applications.

During the past two decades, researchers have established several effective nonlinear optical (NLO) mechanisms for optical limiting, namely, reverse saturable absorption (RSA), two-photon absorption (TPA), free-carrier absorption, nonlinear scattering and nonlinear refraction [5,8]. Fullerene-C₆₀ is considered an ideal optical limiter across the visible spectrum (from 400 to 700 nm) since it was first

studied by Tutt and Kost [14]. The primary mechanism of optical limiting for C₆₀ is RSA, which involves formations of excited states (triplet-triplet) with an absorption cross section much larger than that of the ground state, leading to nonlinear absorption. More recent studies show CNTs have potential as a broadband optical limiter from 430 to 1064 nm [4,7,8,15]. Thermally-induced nonlinear scattering and nonlinear refraction are found to be the dominant mechanisms. At low fluences, nonlinear scattering comes from solvent microbubbles generated by heat transfer from CNTs. At high fluences, the scattering is mainly due to formations of carbon microplasmas. Therefore, the superior optical limiting performance of CNTs can be attributed to their unique electronic structures that absorb broadband energy, as well as their large surface area that can effectively dispatch the energy to surrounding environments. In contrast to Fullerenes with a quick response time in the picosecond regime, CNTs generally respond at best in the nanosecond regime.

The development of nonlinear absorber–CNT hybrids by covalent or noncovalent combination has been reported by several groups [19-23]. In this work, we report the synthesis, characterization and optical limiting property of a covalent linked CNT-Fullerene hybrid. The hybrid is expected to merge complementary temporal and spatial NLO characteristics of Fullerene and carbon nanotubes, resulting in a viable optical limiting material satisfying most of requirements for laser protection.

2. Experimental Methods

Materials. DWNTs prepared using a CVD method (> 95% purity, O.D. of 4 nm and length of 5-20 μm) were obtained from Nanolab. Transmission electron microscope (TEM) images revealed that the pristine DWNT sample is actually a mixture of SWNT, DWNT, TWNT and FWNT. Polyethylenimine (PEI, branched, $M_n = 10,000$ Da), *N,N*-dimethylformamide (DMF, anhydrous, 99.8%), Fullerene-C₆₀ (98%), chlorobenzene (anhydrous, 99.8%) were obtained from Aldrich and used as received.

Direct Amination of DWNTs. DWNTs (3 g) and polyethylenimine ($M_n \sim 10,000$) (15 g) were mixed in 250 mL of DMF. Sonication for 20 min and stirring at 50 °C for 5 d formed the product, DWNT-PEI 10,000. The resulting suspension was filtered through a 0.20 μm nylon membrane and the precipitate

was washed with 1M HCl, 1M NaOH, water and methanol to remove any excess PEI. After drying, 5.5 g of the product was obtained. Characterization of the product was carried out by titration. A known amount of DWNT-PEI10,000 was added to 20 mL of 0.01M HCl. After sonicating the mixture for 1 h at 25 °C, an aliquot (5 mL) of the resulting HCl solution was titrated to a pH 9.0 endpoint using 0.01M NaOH. The difference between the original amount of HCl and the amount of HCl titrated corresponded to the loading of titratable amino groups on DWNT-PEI10,000 and was 0.9 mequiv of titratable amine groups/g.

Synthesis of DWNT-PEI10,000-C₆₀. DWNT-PEI10,000 (300 mg), Fullerene-C₆₀ (150 mg) and triethylamine (0.5 mL) were mixed in 20 mL of chlorobenzene. Sonication for 1 min and stirring at 100 °C for 5 d formed the product, DWNT-PEI10,000-C₆₀. The resulting suspension was filtered through a 0.20 µm nylon membrane and the precipitate was washed with chlorobenzene and methanol to remove any excess C₆₀ and triethylamine. After drying, 408 mg of the product was obtained.

Synthesis of PEI10,000-C₆₀. PEI10,000 (50 mg), Fullerene-C₆₀ (100 mg) and triethylamine (1.0 mL) were mixed in 10 mL of chlorobenzene. Sonication for 1 min and stirring at 100 °C for 5 d formed the product, PEI10,000-C₆₀. The resulting suspension was filtered through a 0.20 µm nylon membrane and the precipitate was washed with chlorobenzene and methanol until the filtrate became colorless. After drying, 122 mg of the product was obtained.

Characterization. Thermogravimetric analysis (TGA) was recorded by SEIKO 1 TG/DTA 200 under Ar or air from 30 to 950 °C. UV-vis spectra were recorded by an Ocean Optics HR2000+ high resolution spectrometer using a Mikropack PH-2000-BAL Deuterium-Halogen light source. High resolution micrographs were taken by an FEI TECNAI G2 20F transmission electron microscope. It was operated at a 200 kV accelerating voltage, equipped with a ZrO₂/W Schottky field emission gun and a Gatan imaging filter slow-scan CCD camera (GIF Tridiem with 2k x 2k pixels). Bright field TEM images were collected by the CCD camera of the TECNAI 20F.

Z-scan Measurement. NLO and optical limiting properties of DWNT-Fullerene hybrids were studied

using open aperture Z-scan method, which is widely adopted to investigate third-order NLO processes, including nonlinear absorption, scattering and refraction. In this work, a Z-scan was performed using 6 ns pulses from a Q-switched Nd:YAG laser. After spatially removing higher-order modes, a laser beam was tightly focused with a 9 cm focal length lens. The laser was operated at its second harmonic, 532 nm, with a repetition rate of 10 Hz. Simultaneously, a focusing lens setup was arranged at $\sim 30^\circ$ to the direct incident beam to monitor the scattered light from the dispersions. All samples were tested in 1.0 cm quartz cells.

3. Results and Discussion

Syntheses and studies of CNT-fullerene hybrids have been reported [24,25]. D'Souza et al. performed syntheses of SWNT/Pyr-NH₃⁺/crown-C₆₀ hybrids by supramolecular assembly and studied their photoinduced electron transfer [24]. The results suggested a possibility of electron transfer from SWNTs to singlet excited Fullerenes in the hybrids. Wu et al. performed syntheses of grapevine SWNT-C₆₀ hybrids by covalent amide formation between oxidized SWNT and C₆₀ derivative [25]. Characterizations revealed a ground-state electron-transfer trend from SWNT to C₆₀ moieties in the grapevine nanostructure. Although these CNT-fullerene hybrids show great potential as materials for nonlinear optics, their syntheses are prohibitively complicated and time consuming. Here, we synthesized CNT-Fullerene hybrids using all commercial available reagents in only two steps. DWNTs are first treated with an excess amount of polyethylenimine (PEI) in *N,N*-dimethylformamide (DMF) at 50°C to give DWNT-PEI10,000 by following a procedure described by Liao et al [26]. Similar procedures were carried out by Dillon et al. for SWNT amination as well [27]. The PEI functionalized DWNTs can readily disperse in common organic solvents and the free amines on PEI functionalized DWNTs can act as nucleophiles. In the next step, DWNT-PEI10,000 was reacted with an excess amount of C₆₀ in chlorobenzene at 100 °C for 5 days, as shown in Figure 1. Amination of Fullerenes was first studied by Hirsch et al [28] and it is suggested that amination of CNTs involves a similar mechanism as amination of Fullerenes [29].

Characterization of CNT-Fullerene hybrids included a series of chemical and physical analyses. Thermogravimetric analysis (TGA) under Ar showed distinct differences between the PEI derivative of DWNTs and starting DWNTs. As shown in Figure 2, as-received DWNTs exhibited a gradual weight loss of 10% over a temperature range 30-950 °C. DWNT-PEI10,000 exhibited a ~ 30% weight loss in the 250-350 °C temperature range. The weight loss in this region was ascribed to loss of the grafted PEI on DWNT samples. The TGA of DWNT-PEI10,000-C₆₀ under Ar exhibited a ~ 25% weight loss at the same temperature range, this represented the loss of grafted PEI. However, no extra weight loss was observed at higher temperature range. Presumably, both DWNT and C₆₀ residue didn't burn under Ar. This hypothesis was proved by a separate TGA of PEI10,000-C₆₀ under Ar over a temperature range 30-800 °C (data not shown). The TGA showed more than 60% of residue was left which can be ascribed to the C₆₀ residue. In order to differentiate the composition of remaining residue of DWNT-PEI10,000-C₆₀, the next TGA was first carried out under Ar from 30 to 500 °C to fully degrade PEI, air was then introduced to the heating chamber and the temperature was increased gradually to 650 °C. A sudden drop of ~ 25% weight was observed at 520 °C, followed by a more gradual weight loss of another 45% from 520 to 600 °C. The first sudden drop was ascribed to the loss of C₆₀ because all C₆₀ molecules should burn at the same temperature. While the gradual weight loss from 520 to 600 °C can be ascribed to the loss of DWNT because it is dependent on the diameter and length of DWNTs. Each individual DWNT should burn gradually under different temperature ranges.

The formation of DWNT-Fullerene hybrids was also confirmed by UV-vis spectroscopy. As shown in Figure 3, relative to the spectrum of C₆₀, the spectrum of DWNT-PEI10,000-C₆₀ in chlorobenzene showed a maximum absorption of 330 nm. The peak at 330 nm can be assigned to the allowed $^1T_{1u} \rightarrow ^1A_g$ transition of C₆₀ [30]. Direct evidence of the formation of nanotube-fullerene hybrids can be seen through high resolution TEM micrographs as well. As shown in Figure 4, while the image of a pristine nanotube (left) shows a smooth surface covered with some amorphous carbon materials, a monomeric form of C₆₀ spheres (right) can be clearly seen to be attached to the inner and outer surfaces

of the SWNT in the image of the nanotube-fullerene hybrid.

For the optical limiting study, the hybrid was dispersed in chlorobenzene and poly(*m*-phenylenevinylene-*co*-2,5-dioctoxyp-phenylenevinylene) (PmPV), respectively. PmPV is a well-known conjugated polymer used to disperse carbon nanotubes, resulting in property modified nanocomposites [3,31]. The normalized transmission and scattered light as functions of incident energy density was shown in Figure 5. As the incident energy density was increased, intense scattered light was observed from DWNT-Fullerene dispersions along with decreasing transmission. Therefore the nonlinear scattering, arising from DWNT moieties, is regarded as one of the main mechanisms for optical limiting. The mechanism leading to optical limiting effects in carbon nanotubes has been studied intensively. The induced scattering centers are composed of solvent bubbles and carbon microplasmas. Due to the solvent effect [8,17], the hybrid dispersed in chlorobenzene possess superior optical limiting response and stronger scattered signal in comparison to the polymer dispersions. In addition, it is believable that RSA from Fullerene moieties has significant contribution as well. The superior optical limiting effect of the hybrids results from an accumulation of different mechanisms.

In order to evaluate the optical limiting ability of DWNT-Fullerene dispersions, we compare the optical limiting response of this hybrid with that of Fullerenes and SWNTs. As shown in Figure 6, the optical limiting performance of this hybrid is superior to those of Fullerenes and CNTs at the same level of transmission. Table 1 summarizes the linear and NLO coefficients for these three materials.

4. Conclusion

In summary, synthesis and characterization of DWNT-Fullerene hybrids was performed to merge complementary temporal and spatial NLO characteristics of Fullerene and carbon nanotubes. The optical limiting performance of DWNT-Fullerene hybrids is superior to those of Fullerenes and SWNTs at the same level of transmission. Future studies regarding mechanisms that contribute to this better optical limiting performance and the influence of covalent linkages between DWNTs and C₆₀ to optical limiting are currently underway.

Acknowledgement

The work is supported by the US Department of Energy under Grant No. DE-FG36-08GO88008 and in part by the Science Foundation Ireland (SFI) under Grant No. 08/CE/I1432. J.W. thanks SFI for his postdoctoral research fellowship.

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Captions for Figures and Table

Figure 1. Synthesis of DWNT-PEI10,000-C₆₀

Figure 2. TGA of DWNT, DWNT-PEI10,000, DWNT-PEI10,000-C₆₀ in Ar and DWNT-PEI10,000-C₆₀ in Ar followed by air introduced at 500 °C

Figure 3. UV-vis spectra of C₆₀, DWNT-PEI10,000 and DWNT-PEI10,000-C₆₀ in chlorobenzene

Figure 4. TEM images of a pristine TWNT partially covered with carbon materials (left) and a SWNT-Fullerene hybrid (right). The circular profiles indicated by the arrows suggest that monomeric C₆₀ molecules (based on their size and shape) are attached to the inner and outer surfaces of the nanotube.

Figure 5. Plots of normalized transmission and scattering light against incident pulse energy density for DWNT-Fullerene dispersions.

Figure 6. Optical limiting responses of the DWNT-Fullerene, Fullerene and SWNTs

Table 1. Linear and NLO coefficients for the DWNT-Fullerene, Fullerene and SWNTs.

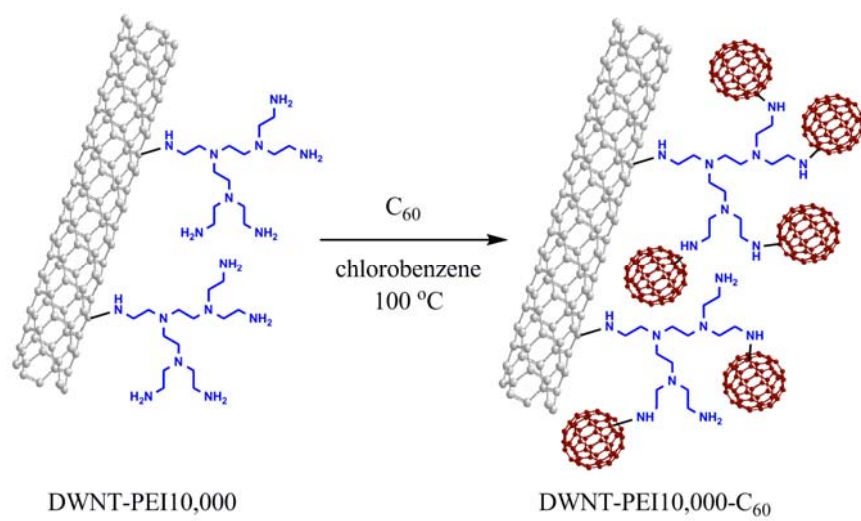


Figure 1.

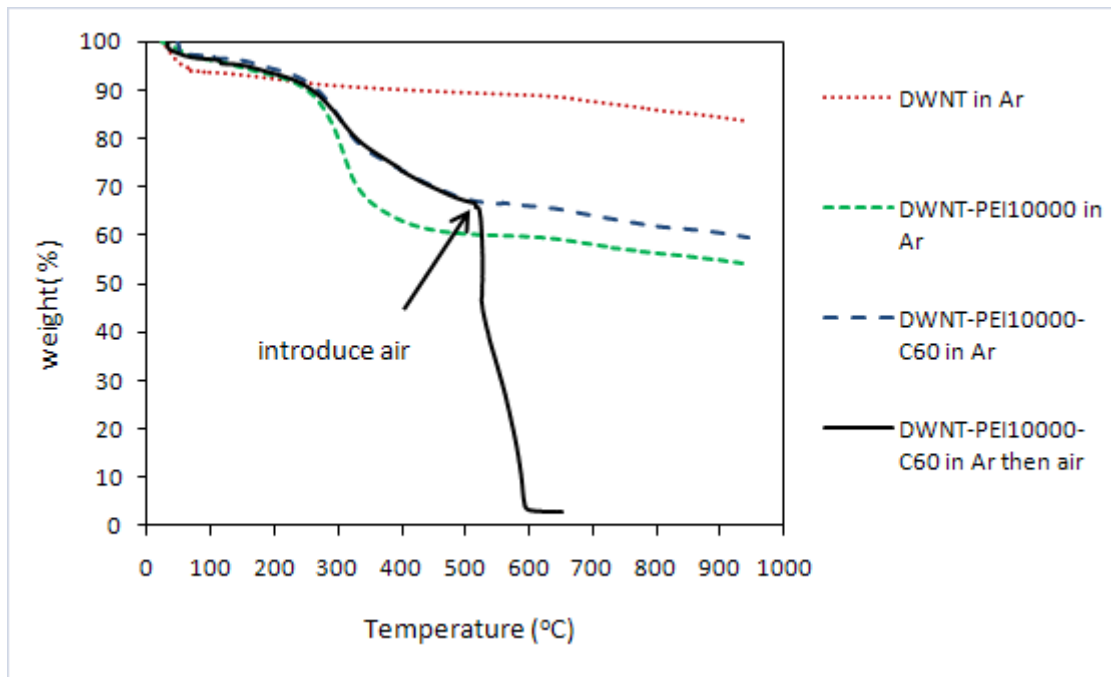


Figure 2.

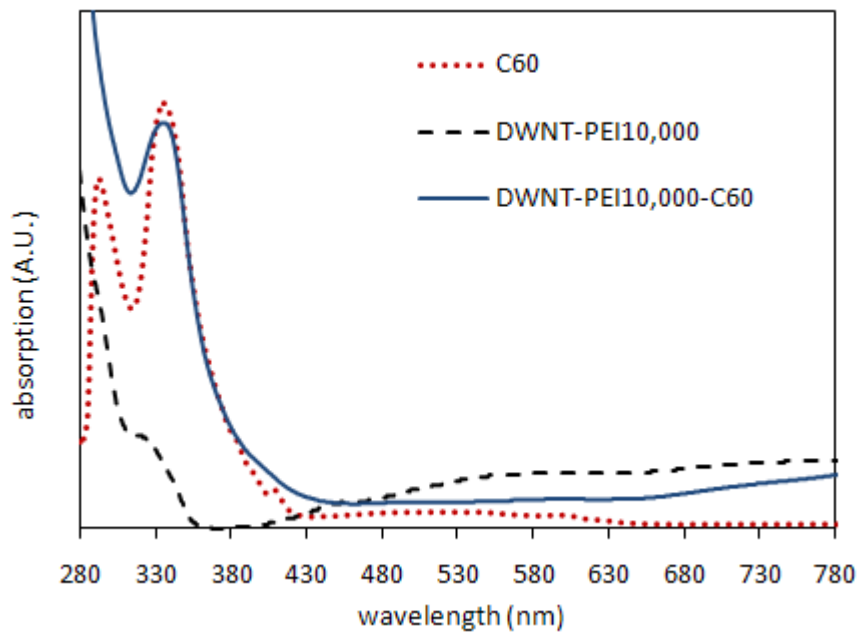


Figure 3.

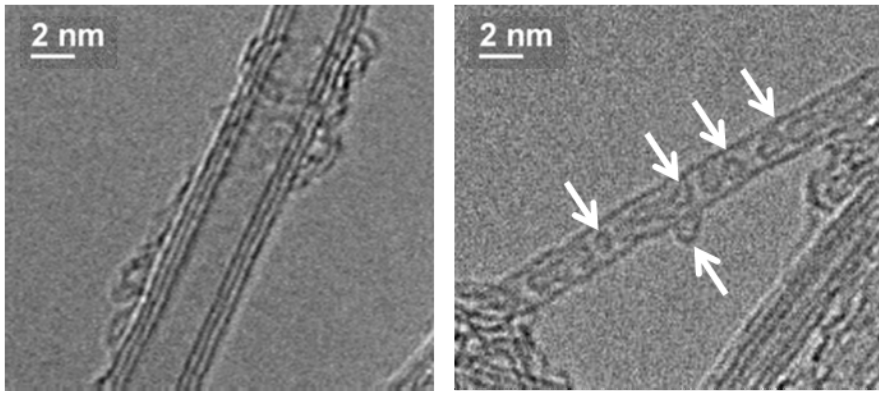


Figure 4.

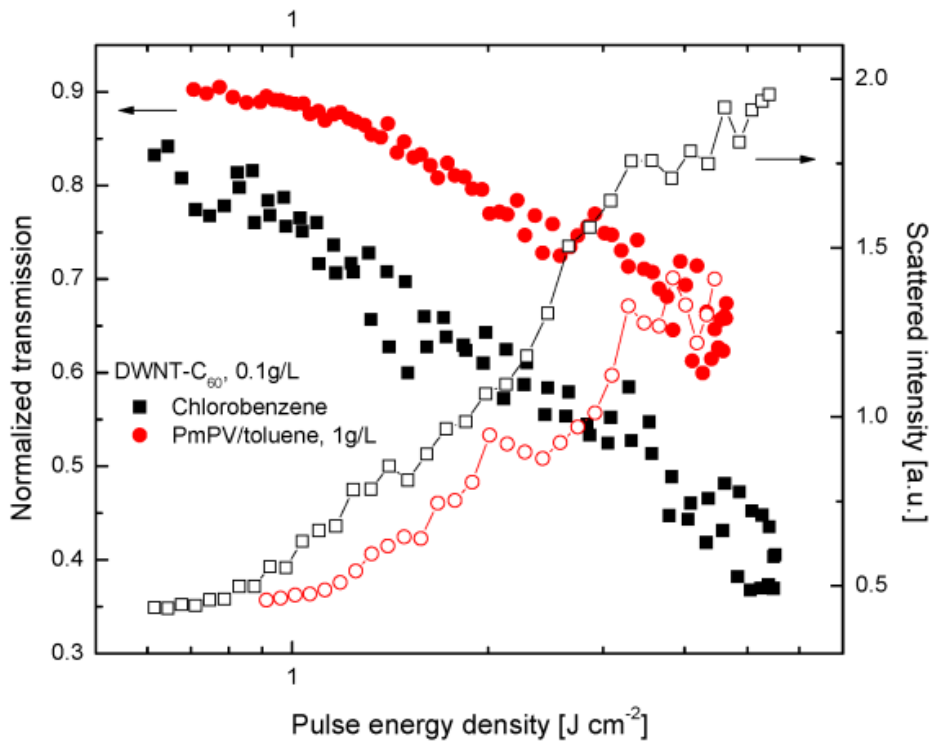


Figure 5.

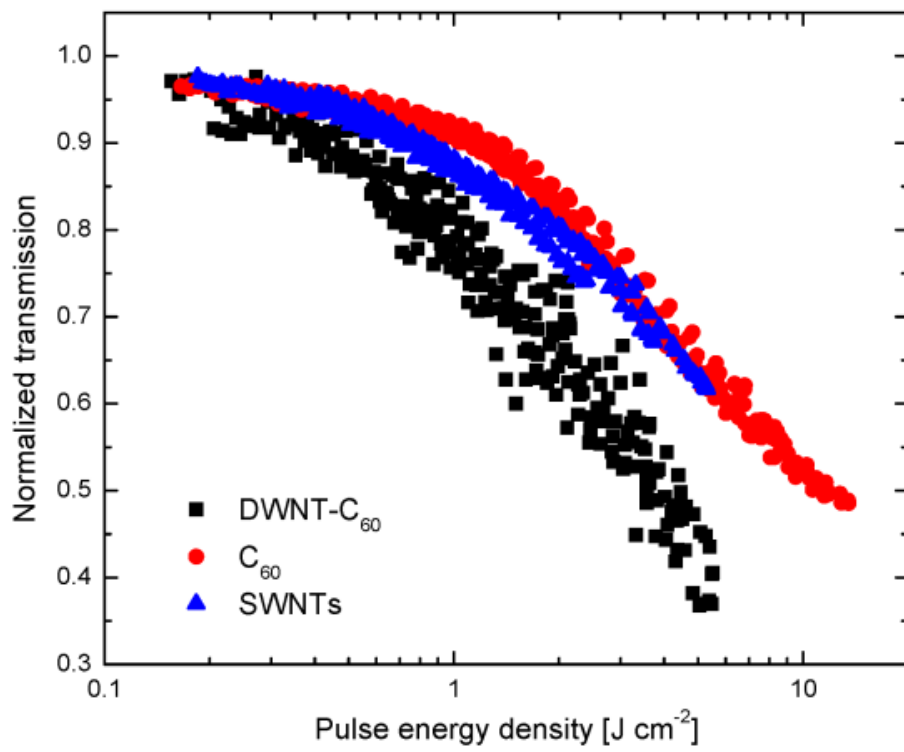


Figure 6.

| Material | Host | Conc. (g/L) | T (%) | α_0 (cm^{-1}) | β_{eff} (cm GW^{-1}) | $\text{Im} \{\chi^{(3)}\}$ ($\times 10^{-12}$ esu) |
|----------------------|---------------|----------------|----------|------------------------------------|---|--|
| DWNT-C ₆₀ | Chlorobenzene | 0.1 | 79.8 | 0.23 | 7.01 ± 0.99 | 2.74 ± 0.39 |
| | PmPV/toluene | 0.1 | 83.4 | 0.18 | 4.01 ± 1.15 | 1.51 ± 0.43 |
| SWNTs | DMF | 0.005 | 81.3 | 0.21 | 2.46 ± 0.50 | 0.85 ± 0.17 |
| C ₆₀ | Toluene | 0.11 | 81.9 | 0.20 | 3.28 ± 0.51 | 1.24 ± 0.19 |

Table 1.