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Optical properties of photopolymerisable nanocomposites containing zeolite nanoparticles

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Acrylamide-based photopolymerisable nanocomposites containing three different types of nanozeolites – Si-MFI, AIPO and BEA, were studied. The influence of the porous nanoparticles on the average refractive index, optical scattering, holographic recording properties and shrinkage were characterised.

The results from the optical characterisation of three different photopolymerisable nanocomposites are presented. Zeolite nanoparticles were chosen as an inorganic additive for improving the dynamic range and shrinkage properties of an acrylamide based photopolymer developed at the Centre for Industrial and Engineering Optics at Dublin Institute of Technology [1-3]. In addition to improving the properties of the holographic recording material [4, 5] the inclusion of a porous nanocomponent and its spatial redistribution during the holographic recording process opens new horizons for the design and fabrication of holographic sensors [6].

The influence of Si-MFI, AIPO-18 and BEA nanozeolites on the nanocomposites optical properties was studied. The zeolite nanoparticles differ in their chemical composition and structure, hydrophobic/hydrophilic nature and pore size [7]. The choice of large-pore size nanoparticles (BEA) [8] and medium to small-pore size nanoparticles (Si-MFI and AIPO-18) [9,10] allows us to distinguish between two situations – when the acrylamide monomer molecules can be trapped inside the nanoparticles (BEA) and when they are larger than the nanoparticles' pore size (Si-MFI, AIPO-18). AIPO-18 nanoparticles are known for their hydrophilic nature and their ability to retain water molecules inside their pores unless heated above given temperature [10] while Si-MFI nanoparticles are understood to remain empty due to their hydrophobic nature [9]. The interaction of the nanoparticles with the host acrylamide photopolymer was studied by Raman spectroscopy of layers prepared from nanocomposites containing the three different types of nanoparticles.

Good compatibility of the zeolite nanoparticles with the host photopolymer was confirmed by Dynamic Light Scattering (DLS) studies performed before and after addition of the nanoparticles to the photopolymer solution. No sign of nanoparticle aggregation in the nanocomposite solution was observed from the DLS results. The concentration of the nanoparticles in the water dispersion did not exceed 5 w/w % but due to the fact that the solvent (water) evaporated during layer preparation the ultimate nanoparticle concentration ranged from 1 to 15 w/w % and from 5 to 50 v/v %. The compatibility of the nanoparticles and the photopolymer components in the dry layers was also confirmed by the low level of optical scattering which was characterised and compared for the three different types of nanoparticles.

The change in the average refractive index in the volume of the nanocomposite layers due to addition of nanozeolites was determined by theoretical modelling of the spectroscopic measurements of their transmission and reflection coefficients. The change in the surface refractive index was studied by the method of the disappearing diffraction pattern using a three wavelength laser refractometer [11].

The influence of the nanoparticle size on the holographic recording properties was studied for Si-MFI nanoparticles with sizes 10, 40 and 60 nm. These results are compared with those obtained using nanocomposites containing 40 nm BEA and 180 nm AIPO-18 nanoparticles. Holographic recording properties are reported for different recording intensities ranging from 1 to 20 mW/cm² and spatial frequencies ranging from 500 to 2000 l/mm.

The spatial redistribution of the nanoparticles as a result of the holographic recording process was confirmed by both SEM-EDX and Raman spectroscopy studies.

It was observed that the addition of zeolite nanoparticle leads to significant suppression of the shrinkage due to polymerisation. The extend of shrinkage was studied by recording slanted transmission holographic gratings and determining the shift of the angular Bragg selectivity curve from its original angular position.

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